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(54) Title: <b>CROSSLINKABLE COATING COMPOSITIONS</b>			
(57) Abstract  An aqueous crosslinkable coating composition comprising i) an autoxidizably crosslinkable polymer, ii) a not autoxidizably crosslinkable vinyl polymer bearing carbonyl groups and iii) carbonyl-reactive groups to crosslink the vinyl polymer.			

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CROSSLINKABLE COATING COMPOSITIONS

The present invention relates to aqueous-based autoxidizably crosslinkable coating compositions, their preparation and use in forming films and coatings.

Autoxidizably crosslinkable polymers are polymers which crosslink on exposure to oxygen. It is known that polymers which contain unsaturated fatty acid residues undergo crosslinking by autoxidation. The unsaturation in polymers from such groups imparts latent crosslinkability so that when a coating composition thereof is dried in air (often in conjunction with a drier salt) the composition undergoes crosslinking, thereby improving its properties such as mechanical properties (improved hardness and durability) and chemical resistance. A further advantage of coatings containing unsaturated fatty acid residues is an improved glossy appearance. EP 379007, EP 0017199 and EP 647655 all describe a one component emulsion which contain autoxidizable polymers with carboxylic acids groups to provide water dispersibility.

However disadvantages of autoxidizably crosslinkable polymers containing unsaturated fatty acid residues are that the curing process is slow (typically taking a period of days) and that yellowing of the coating occurs in time.

It is known in, for example, US 4198811 and WO97/26303 to blend autoxidizably crosslinkable polymers with acrylic polymer dispersions to reduce yellowing and in some cases cost, however the performance of the resulting coatings, in particular the mechanical properties and the resistance to chemicals is greatly reduced.

We have surprisingly found that compositions of autoxidizably crosslinkable polymers containing unsaturated fatty acid residues with a polymer which is not autoxidizably crosslinkable, but has certain other crosslinking means, not only reduces the cure time and yellowing of coatings derived from such a composition, but also that the performance of such a composition in other respects (such as the mechanical properties and chemical resistance) is not vitiated and is comparable to an improvement to the performance of the autoxidizable crosslinkable polymer containing unsaturated fatty acid residues alone.

According to present invention there is provided an aqueous crosslinkable coating composition comprising as aqueous dispersed components:

- i) at least one autoxidizably crosslinkable organic polymer containing unsaturated fatty acid residues, and
- ii) at least one vinyl polymer which is not autoxidizably crosslinkable and bears carbonyl functional groups, and

wherein said composition has present therein carbonyl-reactive amine and/or hydrazine functional groups which impart crosslinkability to component (ii).

For the sake of clarity by a carbonyl functionality in this specification (unless specified otherwise) is meant the carbonyl functionality of a ketone or aldehyde group.

Such carbonyl functional groups in a vinyl polymer are normally chain-pendant and/or terminal groups.

By a hydrazine functional group is meant a carbonyl-reactive functional group of formula  $-NHNH_2$  and also a carbonyl-reactive hydrazone group derived from such a group by reaction with a monoketone or monoaldehyde of at least two carbon atoms. It will become apparent that such functional groups are usually part of larger groups, such as those of formulae,  $-R-C(=O)-NH-NH_2$ ,  $-R-C(=O)-NH-N=C<$ ,  $-R-NH-C(=O)NH-NH_2$ ,  $-R-NH-C(=O)-NH-N=C<$ , and  $-R-NH-NH_2$ , where R is optionally substituted alkylene, optionally substituted alicyclic, or optionally substituted aryl.

By an amine functional group is meant a carbonyl-reactive group of formula  $-NH_2$  and carbonyl-reactive groups derived from such groups. Examples of amine functional groups include  $R-NH_2$ ,  $R-O-NH_2$ ,  $R-O-N=C<$ ,  $-R-NH-C(=O)-O-N=C<$  and  $-R-NH-C(=O)-O-NH_2$  where R is as described above.

By an autoxidizably crosslinkable polymer is meant a polymer which crosslinks on exposure to oxygen from air by virtue of containing unsaturated fatty acid residues.

By a not autoxidizably crosslinkable polymer is meant a polymer not containing unsaturated fatty acid residues.

For the purpose of this invention an "aqueous dispersion" of a polymer(s) means a dispersion of polymer(s) in a liquid carrier medium of which water is the principle component (at least 50 weight %, more usually at least 80 weight %, of the carrier medium). Minor amounts of organic liquids may optionally be present. The dispersion will typically comprise colloidal dispersed particles of the polymer(s), i.e. will typically be in the form of an aqueous latex(ices).

Preferably the autoxidizably crosslinkable organic polymer containing unsaturated fatty acid residues is a polyurethane polymer. Other suitable polymers include alkyds which may be self-emulsifiable, water-based or emulsified alkyds.

In the invention composition, the carbonyl-reactive amine and/or hydrazine functional groups impart crosslinkability to the vinyl polymer(s) of component (ii) by virtue of the latter bearing carbonyl functional groups. It is nevertheless within the scope of the invention, albeit less preferred, for such amine and/or hydrazine functional groups to also impart crosslinkability to the polymer(s) of component (i) if this polymer(s) also bears carbonyl-functional groups - as is possible, but less preferred.

The disposition of the amine and/or hydrazine functional groups in the invention composition to impart crosslinkability as defined may be realised in various ways. More preferably hydrazine functional groups are used in the invention composition.

In a preferred embodiment, the carbonyl-reactive amine and/or hydrazine functional groups are present in the composition by virtue of being provided by added discrete compounds bearing 2 or more amine groups, hereinafter called polyamines, or 2 or more hydrazine functional groups, hereinafter called polyhydrazines. By "discrete" is

meant that such compounds are exclusive of the polymers of components (i) and (ii). Preferably the polyamines or polyhydrazines are diamines or triamines (i.e. having 2 or 3 amine groups) or dihydrazines (i.e. having 2 hydrazine functional groups).

The presence of such a polyamine or polyhydrazine will impart crosslinkability to the vinyl polymer(s) of component (ii). However, if the polymer(s) of component (i) also bears carbonyl functional groups (as is possible as discussed above), such a polyamine or polyhydrazine will also impart crosslinkability to this polymer(s) of component (i) by reaction with the carbonyl functional groups thereof, i.e. in addition to crosslinkability by autoxidisation.

It is, furthermore, also possible for amine and/or hydrazine functional groups to be borne on the polymer(s) of component (i), i.e. the autoxidisably crosslinkable organic polymer containing unsaturated fatty acid residues could also bear carbonyl-reactive amine or hydrazine groups, thereby allowing the organic polymer(s) to undergo crosslinking by reaction with the carbonyl functional groups of the vinyl polymer(s) as well as crosslinking by autoxidation.

As discussed above, the polymer of component (i) in this embodiment may itself also bear carbonyl functional groups, and in such a case crosslinkability could also be provided by reaction between the polymer-based amine or hydrazine groups of the polymer(s) of component (i) and the polymer-based carbonyl functional groups of the polymer(s) of component (i) (in addition to that by reaction with the polymer-based carbonyl functional groups of the vinyl polymer(s) of component (ii)).

In a yet further embodiment, the not autoxidisably crosslinkable vinyl polymer(s) of component (ii) may in addition to bearing carbonyl functional groups also bear carbonyl-reactive amine or hydrazine groups. In such a case, crosslinkability could be provided by reaction between these polymer-based carbonyl functional groups and polymer-based amine or hydrazine groups of the vinyl polymer(s) of component (ii) (if present).

It is apparent from the foregoing, that more than one of such embodiments could be operative in the same invention composition. However, in the most preferred embodiment, the amine and/or hydrazine groups are present only in discrete added polyamines and/or polyhydrazines and the carbonyl functional groups are present only in the vinyl polymer(s) of component (ii). Alternatively a mixture of discrete added and polymer based amine and/or hydrazine groups may be employed.

Autoxidisably crosslinkable polyurethane polymers containing unsaturated fatty acid residues are preferably obtained from the reaction of at least one organic polyisocyanate with at least one isocyanate-reactive organic compound bearing unsaturated fatty acid residue(s), optionally (but preferably) with isocyanate-reactive organic compounds bearing water-dispersing groups, and in some embodiments (as implied above), isocyanate-reactive compounds bearing carbonyl and/or carbonyl-reactive amine or hydrazine groups. Optionally, the reactants may also include a low molecular

weight isocyanate-reactive compound(s) (preferably with a molecular weight of below 500), usually an organic polyol and/or a high molecular isocyanate-reactive compound(s) (preferably with a molecular weight of from 500 to 6000), also usually an organic polyol - such compounds, if used, bearing neither unsaturated fatty acid residue(s) nor water-dispersing groups. Isocyanate-reactive groups include -OH, -NH-, and -NH<sub>2</sub>.

The polyurethane polymer may be prepared in a conventional manner by reacting the organic polyisocyanate(s) with the isocyanate-reactive compound(s) by methods well known in the prior art. Preferably an isocyanate-terminated polyurethane prepolymer is first formed, which is chain extended with an active hydrogen containing compound. If the polymer is made in such manner, the unsaturated fatty acid residue(s) bearing compound is introduced into the polyurethane backbone during the prepolymer formation and/or during the chain extension step. The optional polymer-based carbonyl, amine or hydrazine groups (see supra) may also be introduced during the prepolymer formation and/or during the chain extension step.

Alternatively a polyurethane (preferably of a low molecular weight) may be made by capping an isocyanate-terminated polyurethane with monofunctional isocyanate-reactive compounds or by using an excess of compounds having isocyanate-reactive groups during polymer preparation or a combination of the above preparations may be used.

Optionally (but preferably) monomer(s) bearing non-ionic or ionic water-dispersing or emulsifier groups (or groups that may be subsequently converted thereto) are included in the prepolymer formation to provide the facility of self-dispersability in water of the polyurethane prepolymer and the final autoxidizably crosslinkable polyurethane polymer.

Preferred isocyanate-reactive compounds bearing unsaturated fatty acid residue(s) which may be used in the urethane synthesis may be obtained from the reaction, using techniques known in the art, of a suitable fatty acid with a hydroxyl donor (preferably an alcohol or polyol) or amine donor to provide a fatty acid residue-bearing compound with at least one (preferably at least two) hydroxyl or amine isocyanate-reactive groups.

Preferred fatty acids include fatty acids derived from castor oil, soybean oil, sunflower oil, tallow oil, linseed oil and fatty acids such as linoleic acid, palmitoleic acid, linolenic acid, oleic acid, oleosteric acid, licanic acid, arachidonic acid, ricinoleic acid and/or mixtures thereof.

Suitable polyisocyanates include aliphatic, cycloaliphatic, araliphatic and/or aromatic polyisocyanates. Examples of suitable polyisocyanates include ethylene diisocyanate, 1,6-hexamethylene diisocyanate, isophorone diisocyanate, cyclohexane-1,4-diisocyanate, 4,4'-dicyclohexylmethane diisocyanate, p-xylene diisocyanate, tetramethylxylene diisocyanate, 1,4-phenylene diisocyanate, 2,4-toluene diisocyanate, 2,6-toluene diisocyanate, 4,4'-diphenylmethane diisocyanate, polymethylene polyphenyl

polyisocyanates 2,4'-diphenylmethane diisocyanate, 3(4)-isocyanatomethyl-1-methyl cyclohexyl isocyanate, and 1,5- naphthylene diisocyanate. Mixtures of polyisocyanates can be used and also polyisocyanates which have been modified by the introduction of urethane, allophanate, urea, biuret, carbodiimide, uretonimine, urethdion or isocyanurate residues.

Other isocyanate-reactive organic compounds bearing neither unsaturated fatty acid residues nor water-dispersing groups (see supra) which may be used in the preparation of polyurethanes or polyurethane prepolymers preferably contain at least one (preferably at least two) isocyanate-reactive groups, and are more preferably organic polyols. The organic polyols particularly include diols and triols and mixtures thereof but higher functionality polyols may be used, for examples as minor components in admixture with diols. The polyols may be members of any of the chemical classes of polyols used or proposed to be used in polyurethane formulations. In particular the polyols may be polyesters, polyesteramides, polyethers, polythioethers, polycarbonates, polyacetals, polyolefins or polysiloxanes. Preferred polyol molecular weights are from 250 to 6000, more preferably from 500 to 3000. Low molecular weight organic compounds containing at least one (preferably at least two) isocyanate-reactive groups and having a weight average molecular weight up to 500, preferably in the range of 40 to 250 can also be used. Examples include ethyleneglycol, 1-propanol, and 1,4-cyclohexyldimethanol.

The water-dispersing group content of the polyurethane (if present) may vary within wide limits but should be sufficient to provide the polyurethane with the required degree of water-dispersibility.

Water-dispersing groups are optionally incorporated into the polyurethane by including an isocyanate-reactive and/or isocyanate compound(s) bearing nonionic and/or ionic water-dispersing groups (or groups which may be subsequently converted to such water-dispersing groups) as reactants in the preparation of the polymer or prepolymer.

Typically, ionic water-dispersing groups are anionic salt groups, for example carboxylate, sulphonate and phosphonate salt groups. Examples of such compounds include carboxy group containing diols and triols, for example dihydroxy alkanolic acids. The preferred carboxy containing diol is 2,2-dimethylolpropionic acid or 2,2-dimethylolbutanoic acid. If desired, the carboxy containing diol or triol may be incorporated into a polyester by reaction with a dicarboxylic acid before being incorporated into the prepolymer. The conversion of any acid groups present in the prepolymer to anionic salt groups may be effected by neutralising the said acidic groups before, after or simultaneously with formation of an aqueous dispersion of the prepolymer.

Non-ionic water-dispersing groups are preferably pendant polyoxyalkylene groups, particularly polyoxyethylene groups.

The polyurethane polymer or prepolymer may have a combination of ionic dispersing groups and non-ionic water-dispersing groups.

A carbonyl functional group may be introduced onto a polyurethane backbone by using for example carbonyl functional compounds with at least one and preferably two isocyanate-reactive groups, for example dihydroxyketones or the adduct obtained by a Michael addition reaction of diacetonacrylamide with a diamine or an alkanol amine.

An amine or hydrazine functional group may similarly be introduced onto a polyurethane backbone. For example a compound is used in which the hydrazine functionality has been blocked by reaction with a monoaldehyde or a monoketone to form a hydrazone structure. This expedient is employed because a hydrazine functionality will otherwise react during the prepolymer preparation and so be partially or even completely removed.

Examples of suitable isocyanate-reactive monoaldehyde or monoketone hydrazine compounds include gammahydroxy butyric hydrazide and semi carbazide ethyl methacrylate blocked with a monoaldehyde or monoketone.

Active hydrogen-containing chain extending compounds which provide hydrazine (or hydrazone) functionality pendant to the polyurethane chain include diamino hydrazides. Such compounds may be prepared by reacting a diamine with an acrylic acid derivative (for example ethyl acrylate) and then reacting the product obtained with hydrazine.

When an isocyanate-terminated polyurethane prepolymer is prepared, it is conventionally formed by reacting a stoichiometric excess of the organic polyisocyanate with the isocyanate-reactive compounds under substantially anhydrous conditions at a temperature between about 30°C and about 130°C until reaction between the isocyanate groups and the isocyanate-reactive groups is substantially complete; the reactants for the prepolymer are generally used in proportions corresponding to a ratio of isocyanate groups to isocyanate-reactive groups of from about 1.1:1 to about 6:1, preferably from about 1.5:1 to 3:1.

If desired, catalysts such as dibutyltin dilaurate and stannous octoate, zirconium or titanium based catalysts may be used to assist prepolymer formation. An organic solvent may optionally be added before or after prepolymer formation to control the viscosity. Examples of solvents include water-miscible solvents such as N-methylpyrrolidone, dimethyl sulphoxide and dialkyl ethers of glycol acetates or mixtures of N-methylpyrrolidone and methyl ethyl ketone. Optionally no organic solvents are added.

An aqueous polyurethane dispersion may then be prepared by dispersing the isocyanate-terminated polyurethane prepolymer (optionally carried in an organic solvent medium) in an aqueous medium (using surfactants, or more preferably by utilising the self-dispersability of the prepolymer if dispersing groups are present therein, although surfactants may still be employed if desired) and chain extending the prepolymer with active hydrogen-containing chain extender in the aqueous phase.



The aqueous polyurethane dispersion may of course comprise more than one polyurethane polymer prepared as defined above.

The urethane prepolymer may be dispersed in water using techniques well known in the art. Preferably, the prepolymer is added to the water with agitation or, alternatively, water may be stirred into the prepolymer.

Active hydrogen-containing chain extenders which may be reacted with the prepolymer include polyol(s), amino-alcohol(s), a primary or secondary diamine(s) or polyamine(s), hydrazine(s) or a substituted hydrazine(s).

Examples of such chain extenders useful herein include alkyene diamines such as ethylene diamine and cyclic amines such as isophorone diamine. Also materials such as hydrazine, azines such as acetone azine, substituted hydrazines such as, for example, dimethyl hydrazine, 1,6-hexamethylene-bis-hydrazine, carbodihydrazine, hydrazides of dicarboxylic acids and sulphonic acids such as adipic acid mono- or dihydrazide, oxalic acid dihydrazide, isophthalic acid dihydrazide, hydrazides made by reacting lactones with hydrazine such as gammahydroxylbutyric hydrazide, bis-semi-carbazine, and bis-hydrazide carbonic esters of glycols may be useful. Water itself may be effective as an indirect chain extender.

Where the chain extender is other than water, for example a polyol, polyamine or hydrazine, it may be added to the aqueous dispersion of prepolymer or, alternatively, it may already be present in the aqueous medium when the prepolymer is dispersed therein. The prepolymer may also be chain extended to form the polyurethane polymer while dissolved in organic solvent (usually acetone) followed by the addition of water to the polymer solution until water becomes the continuous phase and the subsequent removal of the solvent by distillation to form an aqueous dispersion.

The chain extension can be conducted at elevated, reduced or ambient temperatures. Convenient temperatures are from about 5°C to 95°C or, more preferably, from about 10°C to 60°C.

The total amount of chain extender materials employed (apart from water) should be such that the ratio of active hydrogens in the chain extender(s) to isocyanate groups in the prepolymer preferably being in the range from 0.1:1 to 2.0:1 more preferably 0.80:1 to 1.7:1.

The not autoxidizably crosslinkable vinyl polymers bearing carbonyl functional groups of component (ii) may be formed by the free-radical addition polymerisation of at least one carbonyl-containing monoethylenically unsaturated monomer with at least one other olefinically unsaturated monomer not providing carbonyl functionality.

Examples of unsaturated monomers which bear carbonyl functional groups include acrolein, methacrolein, diacetone-acrylamide, crotonaldehyde, 4-vinylbenzaldehyde, vinyl alkyl ketones of 4 to 7 carbon atoms such as vinyl methyl ketone, and acryloxy- and methacryloxy-alkyl propanols. Further examples include

acrylamidopivalaldehyde, methacrylamidopivalaldehyde, 3-acrylamidomethyl-anisaldehyde, diacetone acrylate, and diacetone methacrylate.

The proportion of carbonyl functional groups in the vinyl polymer is preferably 3 to 200 milliequivalents per 100 g polymer, more preferably 6 to 100 milliequivalents per 100 g polymer.

Examples of non-carbonyl-providing olefinically unsaturated monomers include 1,3-butadiene, isoprene, styrene, divinyl benzene, acrylonitrile, methacrylonitrile, vinyl halides (such as vinyl chloride), vinyl esters (such as vinyl acetate, vinyl propionate and vinyl laurate), hetero cyclic vinyl compounds, alkyl esters of monolefinically unsaturated dicarboxylic acids (such as di-n-butyl maleate and di-n-butyl fumarate) and, in particular, esters, of acrylic acid and methacrylic acid, examples of which are methyl acrylate, methyl methacrylate, ethyl acrylate, ethyl methacrylate, n-butyl acrylate, n-butyl methacrylate, 2-ethylhexyl acrylate, 2-ethylhexyl methacrylate, isopropyl acrylate, hydroxyethyl methacrylate, hydroxypropyl methacrylate, isopropyl methacrylate, n-propyl acrylate and n-propyl methacrylate.

Olefinically unsaturated monomers bearing anionic water-dispersing groups may also be used, examples of which include acrylic acid, methacrylic acid, itaconic acid and/or maleic acid.

Olefinically unsaturated monomers having non-ionic groups, such as alkoxypolyethylene glycol methacrylates, may also be used.

The not autoxidizably crosslinkable vinyl polymer may bear in addition to the carbonyl functional groups, amine and/or hydrazine functional groups as mentioned above (although this is not preferred). Hydrazine functional groups may preferably be introduced by polymerising into the vinyl polymer at least one olefinically unsaturated monomer with (chain-pendant) hydrazinolysable groups which are subsequently reacted with a hydrazine yielding agent to convert at least a proportion of the hydrazinolysable groups into hydrazine functional groups.

Examples of monomers providing chain-pendant hydrazinolysable groups include  $\alpha$ -chloroacrylic acid and especially acid chlorides or esters of acrylic acid, and also acid chlorides or esters of methacrylic acid. Preferred methacrylic acid esters are methyl, ethyl, propyl, isopropyl, n-butyl, tertiary or secondary butyl esters, most preferred are methyl and ethyl esters.

The hydrazine content (if present) of the vinyl polymer is typically up to 300 milliequivalents per 100g polymer, preferably 10 to 200 milliequivalents per 100g polymer.

Hydrazinolysis may be effected by dissolving or dispersing the vinyl polymer in a water-miscible alcohol or a water-alcohol mixture followed by the addition of hydrazine or hydrazine hydrate.

A vinyl polymer bearing carbonyl functionality may be prepared by any suitable free-radical initiated polymerisation technique, a free-radical initiator and appropriate

heating (e.g. 40°C to 90°C) being employed. The polymerisation is normally effected in an aqueous medium, and in particular aqueous emulsion polymerisation is used to prepare the polymer with optionally conventional dispersants being used. Free radical initiators include hydrogen peroxide, t-butylhydroperoxide, persulphates such as NH<sub>4</sub> persulphate K persulphate and Na persulphate or a redox system may be used.

Not autoxidizably crosslinkable vinyl polymers bearing carbonyl functional groups for use in the invention preferably have a weight average molecular weight within the range 1000 to 5,000,000 (more usually 2,000 to 1,000,000). The weight average molecular weight may be measured by gel permeation chromatography (gpc).

Preferably polyamines or polyhydrazines used to crosslink carbonyl functional groups present in the composition are incorporated into components (i) and/or (ii) before mixing the components, or after or during the mixing of components (i) and (ii), as discrete entities.

Preferred polyamines include but are not limited to those with primary and/or secondary amino groups having from 2 to 10 such amino groups per molecule. Especially preferred are primary amines. Suitable examples include ethylenediamine, 4-amino-1,8-octanediamino propylenediamine, decamethylene diamine, 1,2-diaminocyclohexane, isophoronediamine, urea, N-(2-hydroxyethyl)ethylenediamine, tris(2-aminoethyl)amine, melamine, diethylenetriamine, dipropylenetriamine, dibutylenetriamine, polyethylene imines and Jeffamines (polyoxyethylene amines available from Huntsman Corporation, Houston, Texas).

Preferred polyhydrazines include but are not limited to dicarboxylic acid bis-hydrazides, bis-hydrazones, specific examples being oxalic acid dihydrazide, malonic acid dihydrazide, succinic acid dihydrazide, adipic acid dihydrazide, and sebacic acid dihydrazide, cyclohexane dicarboxylic acid bis-hydrazides, azelaic acid bis-hydrazides; also carbonic acid hydrazides, bis-semicarbazides, trihydrazides, dihydrazinoalkones and dihydrazines of aromatic hydrocarbons, for example 1,4-dihydrazinobenzene and 2,3-dihydrazinonophthalene, dihydrazine and others known in the art. Preferred examples include adipic acid dihydrazide and carbonic acid dihydrazides.

If discrete polyamines or polyhydrazines are added to the composition, the level is preferably that to provide a range of 0.02 to 1.6 moles more preferably 0.05 to 0.9 moles of amine or hydrazine groups present per mole of carbonyl functional groups present in the composition.

Component (ii), in addition to the at least one not autoxidizably crosslinkable vinyl polymer bearing carbonyl functional groups optionally comprises a further not autoxidizably crosslinkable polymer or polymer moiety. Suitable not autoxidizably crosslinkable polymers include vinyl polymers not bearing carbonyl functional groups, polyurethane polymers not bearing carbonyl functional groups and/or polyurethane polymers bearing carbonyl functional groups.

Preferably the additional not autoxidisable crosslinkable polymer is a polyurethane polymer or polyurethane polymer moiety. If a not autoxidisable crosslinkable polyurethane polymer is included in the composition as part of component (ii) (see later) it may be made in the same manner as the autoxidisable crosslinkable polyurethane as described above but without the introduction of unsaturated fatty acid bearing compounds.

The additional not autoxidisable crosslinkable polymer or polymer moiety is a) optionally a discrete polymer (i.e. not covalently bonded to the not autoxidisable crosslinkable vinyl polymer bearing carbonyl functional groups), b) formed by *in-situ* polymerisation in the presence of the at least one not autoxidisable crosslinkable vinyl polymer bearing carbonyl functional groups. Alternatively the not autoxidisable crosslinkable vinyl polymer bearing carbonyl functional groups is formed by *in-situ* polymerisation in the presence of the additional polymer.

Alternatively the additional polymer may be grafted to the not autoxidisable crosslinkable vinyl polymer bearing carbonyl functional groups or may comprise a copolymerised moiety of the not autoxidisable crosslinkable vinyl polymer bearing carbonyl functional groups.

The autoxidisable crosslinkable polymer of component (i) and the not autoxidisable crosslinkable polymer(s) of component (ii) of the composition of the invention may be brought together by any suitable technique.

For example an aqueous dispersion of component (i) and an aqueous dispersion of component (ii) may be mixed together with agitation. The aqueous solution may also contain water-miscible organic solvents.

Alternatively the not autoxidisable crosslinkable vinyl polymer may be prepared in the presence of the autoxidisable crosslinkable polymer (component (i)). For example free-radical initiators may be added to a dispersed mixture of component (i) and vinyl monomer(s) or *vice versa* and polymerisation to the vinyl polymer effected. Optionally the not autoxidisable crosslinkable vinyl polymer may be grafted to the autoxidisable crosslinkable polymer.

The weight ratio of organic polymer(s) of component (i) to the not autoxidisable crosslinkable vinyl polymer(s) bearing carbonyl functional groups in the composition is suitably in the range from 90 : 10 to 10 : 90 more preferably of from 70 : 30 to 30 : 70.

Preferred pH ranges are 4 to 11, more preferably 6.5 to 9.5 and especially 7 to 8.5.

Neutralisation agents which may be applied include organic bases, for example hydroxides of lithium, sodium or potassium, and organic bases for example ammonia or tertiary amines for example dimethyl ethanol amine and triethylamine morpholine.

It is preferred to store the invention composition in a closed container, where due to the low amount of oxygen no significant crosslinking of the autoxidisable polymer has

been found to occur until after application. However, this is by no means always necessary and indeed it is possible to use the invention composition soon after its production. After coating the composition the water evaporates, oxygen in the atmosphere initiates autoxidation and a dual crosslinking reaction takes place.

5 The aqueous composition of the invention may be advantageously employed as coating compositions, (e.g. protective or adhesive coating compositions) for which purpose they may be further diluted with water and/or organic solvents, or they may be supplied in more concentrated form by evaporation of water and/or organic components of the liquid medium. As coating compositions, they may be applied to any substrate  
10 including wood, metals, glass, cloth, leather, paper, plastics, foam and the like, by any conventional method including brushing, dipping, flow coating, spraying, and the like.

Drier salts preferably comprise part of the composition. Examples include polyvalent salts containing cobalt, calcium, copper, zinc, iron, zirconium and manganese as the cation and halides, nitrates, sulphates, acetates, naphthenates or acetoacetonates as the anion. The amount of drier used is in the range from 0 to 1% metal content by  
15 weight of the composition.

The compositions may contain other conventional ingredients including organic solvents, pigments, dyes, emulsifiers, surfactants, thickeners, heat stabilisers, levelling agents, anti-cratering agents, fillers, sedimentation inhibitors, UV absorbers, antioxidants, waxes and the like introduced at any stage of the production process or subsequently. It  
20 is possible to include an amount of an antimony oxide in the dispersions to enhance the fire retardant properties. The dispersions may also be used as adhesives for materials such as polypropylene, polyester, polyurethane, leather and the like or as binding agents for various particulate materials.

25 The dispersions suitably have a solids contents of from about 20 to 60% by weight, preferably from about 25 to 45% by weight.

If desired, the compositions of the present invention may include other polymer dispersions for example polyurethane, polyamide, polyepoxide, polyvinyl acetate, polyethylene, polystyrene, polybutadiene, polyvinyl chloride, polyacrylate and other  
30 homopolymer and copolymer dispersions. These can sometimes be prepared in-situ (for example by polymerisation of the monomers in the presence of the polyurethane polymer or prepolymer).

The weight ratio of the organic polymer of component (i) and component (ii), the not autoxidisably crosslinkable vinyl polymer(s) bearing carbonyl functional groups to other included polymers in the composition is preferably from 100:0 to 40:60, more  
35 preferably 100:0 to 70:30, most preferably 100:0.

There is still further provided according to the invention a coated substrate having a coating obtainable or derived from an aqueous crosslinkable coating composition

comprising a dispersion as defined above, and the use of such a composition for coating a substrate.

The present invention is now illustrated by the following examples. Unless otherwise specified, all parts and percentages are on a weight basis.

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Preparation of autoxidisably crosslinkable organic polymers containing unsaturated fatty acid residues = FATTY ACIDS 1 to 5

FATTY ACID 1

10

An unsaturated fatty acid residue containing urethane dispersion

Stage 1 - Synthesis of a fatty acid residue containing polyester polyol for incorporation into a polyurethane

Components	Ingredients	parts by weight
1	Adipic acid	535.9
2	Prifac 8960 (a polyunsaturated fatty acid from Unichema)	916.0
3	1,4-Cyclohexane dimethanol	250.5
4	Trimethylolpropane	411.7
5	Fascet 2005 (SnCl <sub>2</sub> )	0.1
6	Cardura E10 (Neodecanoic acid glycidyl ester)	97.7
7	N,N- dimethylbenzylamine	1.9

15

Components 1, 2, 3 and 4 were charged to a 2.5 litre flask fitted with a stirrer, condenser, nitrogen inlet and thermocouple. The resultant mixture was gradually heated to the reaction temperature of 215 °C. Water of distillation was collected. At this reaction temperature component 5 was added and the mixture was esterified until an acid value of 10.4 mg KOH/g was obtained. Subsequently the temperature was reduced to 170°C and components 6 and 7 were added and reacted until an acid value of 0.1 mg KOH/g was obtained. The total reaction time was 12 hours. The hydroxyl value of the polyester polyol was 76.8 mg KOH/g.

20

Stage 2 - Synthesis of a fatty acid residue containing polyurethane prepolymer

An isocyanate terminated urethane prepolymer containing the polyester polyol prepared in Stage 1 was prepared as described below:

269.6g of Desmodur W (4,4'-disisocyanatodiphenylmethane diisocyanate), 34.0g of a difunctional isocyanate-reactive non-ionic component with a joint hydroxy number and acid number of 87 and a propylene oxide/ethylene oxide (PO/EO) ratio of 9/1, 34.0 g dimethylol propanoic acid (DMPA), 342.4 g of the polyester polyol from stage 1 and 120.0g of N-methyl pyrrolidone (NMP) were added to a 1 litre 3 necked round bottom flask. The mixture was stirred under nitrogen. The mixture was heated to 55°C and 4 drops of Dabco T-9 (a tin octoate catalyst from Air Products) were added. The mixture was then heated to 95°C and kept at this temperature for 1 hour. 4 Drops of Dabco T-9 were added and the mixture was kept at 95°C for another hour. 8.7g sample was taken for isocyanate (-NCO) determination. The residual NCO content was 5.05%. The batch was cooled down to 75°C.

25.6g of triethylamine (TEA) and 10.2 g of Dapro 5005 (manganese drier salt mixture) were added to the prepolymer. The mixture was kept at this temperature for 40 minutes.

Stage 3 - Preparation of a polyurethane dispersion from the polyurethane prepolymer prepared in Stage 2.

600.0 g of the prepolymer prepared in Stage 2 was added to a waterphase consisting of 820.7 g demineralised water and 19.6 g of Atlas G-4809 a nonionic alkoxyated alkylphenol surfactant (70%). The total addition time was 65 minutes. The prepolymer was kept at 70-75°C during the addition to the waterphase, the temperature of the waterphase was 30-32°C.

When the addition was complete, a mixture of 15.6 g of hydrazine hydrate (63.7%) and 25 g of demineralised water was added to the dispersion. The temperature of the dispersion increased to 40°C and was kept at this temperature for 30 minutes. The batch was cooled to room temperature and sieved through a 200 mesh sieving cloth.

The dispersion had a solids content of 35%, pH of 8.1, viscosity of 210 mPas at 25°C and a sediment content of <0.05%.

FATTY ACID 2A fatty acid residue containing low molecular weight (non chain extended) polyurethaneStage 1 - Preparation of amide ester diol products from soybean oil.

A reactor was charged with 148.8g diethanolamine and 1.2g sodium methoxide and the mixture was heated to 90-100°C until all of the sodium methoxide had dissolved.

Subsequently, 750g soyabean oil (available from Alnor Oil company) was added to the mixture followed by stirring and heating to 105-110°C for 4.5 hours.

Gel Permeation Chromatography (GPC) of the amide ester diol product showed its composition to consist of 10.7% unreacted soyabean oil, 41.5% diglyceride mono-ol and 47.8% monoglyceride amide diol (ester amide polyol). The theoretical hydroxy equivalent weight of the amide ester diol product was 211 and corresponding OH number was 266 mg KOH/g.

#### Stage 2 - Preparation of a uralkyd resin

116.3g of the amide ester diol product from Stage 1 was mixed in a reactor with 18g 2,2-dimethylolpropionic acid, 12.8g cyclohexanedimethanol, 20.3g methoxypolyethyleneoxide glycol (MPEG750), 134.2 g N-methylpyrrolidone, and 118.3g toluene diisocyanate. The temperature was maintained at 50-60°C for one hour. Then more amide ester diol product (133.7g) was added and the temperature increased to 70-80°C for two hours.

#### Stage 3 - Preparation of an aqueous uralkyd resin dispersion

The product from Stage 2 was diluted with 76.7g dipropylene glycol monomethyl ether (acting as a cosolvent to reduce viscosity) and 9.55g dimethylethanolamine dissolved in 187g water (base for neutralising carboxyl groups). 5.6 g Drier salt (Dapro 5005, available from Daniels Products) followed by 520 g water at 60°C was added to the mixture while stirring.

The resulting dispersion had a clear, transparent yellow colour with a viscosity of 1640 mPa.s at 25°C, a pH of 7.1 and solids content of 31%.

#### FATTY ACID 3

##### An autoxidisable crosslinkable polyester containing unsaturated fatty acid residues

Worléesol 31A (available from Worlée-Chemie GmbH) is an anionically stabilised polyester alkyd resin containing unsaturated fatty acid residues, neutralised with ammonia and is supplied as a solution in a water/butylglycol mixture (80/20). The resin has a solid content of 45.9%, a viscosity at 20°C of 127 s (according to DIN 53211), the acid number on solids is 100.3 mg KOH/g and pH is 7.8. This solution was used as supplied.

#### FATTY ACID 4

##### An autoxidisable crosslinkable urethane dispersion containing unsaturated fatty acid residues

NeoRez R2001 (available from NeoResins, Avedica BV) is a chain extended urethane dispersion containing unsaturated fatty acid residues. (NeoRez and NeoResins are Trade Marks of Avedica BV). The fatty acid content is approximately 35% on solids. The dispersion has a solids content of 34.9%, pH was 7.8 and the viscosity was 220 mPa.s. This dispersion was used as supplied.



#### FATTY ACID 5

##### An autoxidisable crosslinkable urethane-acrylic hybrid dispersion containing unsaturated fatty acid residues

- 5 XR9405 (available from Avecia Inc) is a urethane-acrylic hybrid dispersion containing unsaturated fatty acid residues. The fatty acid content is approximately 40% on solids. The dispersion has a solids content of 30%, pH was 7.6 and the viscosity was 212 mPa.s. This dispersion was used as supplied.

- 10 Synthesis of not autoxidisable crosslinkable vinyl dispersions bearing carbonyl functional groups = NAVP's 6 to 9

#### NAVP 6

##### A not autoxidisable crosslinkable vinyl dispersion bearing carbonyl functional groups

- 15 912 g of demineralised water and 1.82 g of sodium lauryl sulphate, (30 w/w) were added to a three necked round bottom flask, equipped with a thermometer, stirrer, cooler and a N<sub>2</sub> inlet. The mixture was heated to 70°C under N<sub>2</sub> atmosphere.

- 5% of a monomer mixture consisting of 188.2 g of demineralised water, 5.45 g of sodium lauryl sulphate (30% w/w), 6.98 g of lauryl mercaptane, 3.49 g of mercapto propionic acid, 43.61 g of methacrylic acid, 34.88 g of diacetone acrylamide and 357.56 g of methylmethacrylate was added to the reactor. The reactor content was then heated to 80°C.

- 20 30% of a solution of ammonium persulphate (87.2g, 1.5% w/w) in water was added to the reactor. The reactor content was heated to 85°C and mixed at this temperature for 5 minutes.

- 25 The remainder of the monomer mixture was added to the reactor after 60 minutes, and the remainder of the ammonium persulphate solution was added to the reactor after 70 minutes. The reactor content was kept at 85°C during this period. After the addition was complete, the reactor content was held at 85°C for another 30 minutes.

- 30 A solution of 34.50 g of ammonia (25% w/w) and 124.4 g of demineralised water was added to the reactor. The reactor content was held at 85°C for another 20 minutes. After this the batch was cooled down to room temperature and sieved through a 200 mesh sieve cloth. The pH was 7.8 and the viscosity was 19 mPas at 25°C.

- 1034.5g of this polymer dispersion was mixed with 195.38 g of demineralised water for 15 minutes at room temperature. 0.22 g of dimethylethanol amine, 157.51 g of butylmethacrylate, and 48.58 g of butylacrylate and 6.38g of diacetone acrylamide were added to the reactor, followed by 1.84 g of tert.-butylhydroperoxide (30% solution in water w/w) and 2.80 g of a 1% (w/w) solution of iron ethylenediaminetetraacetate (FeEDTA) in water. The reactor content was heated to 35°C and mixed for 60 minutes.

2.69g of a 3.96 % (w/w) solution of L-ascorbic acid in water was added and the reactor content was heated to 57°C in 25 minutes. The batch was held at this temperature for 15 minutes.

Then 36.98 g of a 1.21% (w/w) solution of L-ascorbic acid in water was added to the reactor in 30 minutes at 57°C. The batch was mixed for 15 minutes and cooled down to 35°C.

A mixture of dimethyl ethanol amine (0.22 g), butyl methacrylate (157.5 g) diacetone acrylamide (6.4g) and butyl acrylate (48.6 g) was added to the reactor. 1.84 g of tert-butylhydroperoxide (30% solution in water w/w), 2.80g of a 1% (w/w) solution of FeEDTA in water and 230.95g of demineralised water were added to the reactor and mixed for 60 minutes at 35°C. 2.69 g of a 3.96% (w/w) solution of L-ascorbic acid in water was added to the reactor content. The batch was heated to 42°C and mixed at this temperature for 15 minutes.

Then 36.98 g of a 1.21% (w/w) solution of L-ascorbic acid in water was added to the reactor in 30 minutes at 42°C.

The batch was mixed for 30 minutes at 40°C and a solution of 15.20g of adipic dihydrazide and 14.13g of demineralised water was added before mixing for a further 30 minutes at 40°C, cooled down to room temperature and sieved through a 200 mesh sieve cloth.

The specifications of this batch were as follows: solids content 35%, pH of 7.8 and viscosity of 18 mPas at 25°C.

#### NVP.7

##### A not autoxidisable crosslinkable acrylic dispersion bearing carbonyl functional groups

859.8 g of demineralised water, 14.0 g of sodium lauryl sulphate (30w/w), 1.69 g ammonium persulphate and 1.03 g of ammonium bicarbonate (25% w/w) were added to a three necked round bottom flask, equipped with a thermometer, stirrer, cooler and a N<sub>2</sub> inlet. 10% of a monomer mixture consisting of 242.0 g of demineralised water, 28.1 g of sodium lauryl sulphate (30% w/w), 2.53 g ammonium persulphate, 2 g ammonium bicarbonate (25% w/w), 50.5 g diacetone acrylamide, 33.7 g methacrylic acid, 421.2 g butylmethacrylate, 157.8 g butylacrylate and 179.1 g styrene was added to the reactor.

This mixture was heated to 80°C under N<sub>2</sub> atmosphere and mixed at this temperature for 20 minutes. The remainder of the monomer mixture was added to the reactor over a period of 90 minutes. The reactor content was kept at 80°C during this period. After the addition was complete, the reactor content was held at 80°C for another 30 minutes. A solution of 3.08 g of ammonia (25% w/w) and 3.34 g of demineralised water was added to the reactor. The reactor content was held at 80°C for another 30 minutes. After this the batch was cooled down to room temperature and sieved through a 200 mesh sieve cloth.

The pH was 6.7 and the viscosity was 400 mPa.s at 25°C. To 1000 g of this dispersion adipic acid dihydrazide (9.1g) was added as a crosslinker.

#### NAVP 8

5 A not autoxidisable crosslinkable acrylic dispersion bearing carbonyl functional groups

859.8 g of demineralised water, 14.0 g of sodium lauryl sulphate (30w/w), 1.69 g ammonium persulphate and 1.03 g of ammonium bicarbonate (25% w/w) were added to a three necked round bottom flask, equipped with a thermometer, stirrer, cooler and a N<sub>2</sub> inlet. 10% of a monomer mixture consisting of 242.0 g of demineralised water, 28.1 g of sodium lauryl sulphate (30% w/w), 2.53 g ammonium persulphate, 2 g ammonium bicarbonate (25% w/w), 25.3 g diacetone acrylamide, 33.7 g methacrylic acid, 421.2 g butylmethacrylate, 162 g butylacrylate and 200.3 g styrene was added to the reactor. This mixture was heated to 80°C under N<sub>2</sub> atmosphere and mixed at this temperature for 20 minutes. The remainder of the monomer mixture was added to the reactor over a period of 90 minutes. The reactor content was kept at 80°C during this period. After the addition was complete, the reactor content was held at 80°C for another 30 minutes. A solution of 3.08 g of ammonia (25% w/w) and 3.34 g of demineralised water was added to the reactor. The reactor content was held at 80°C for another 30 minutes. After this the batch was cooled down to room temperature and sieved through a 200 mesh sieve cloth. The pH was 6.5 and the viscosity was 49 mPa.s at 25°C. To 1000 g of this dispersion 4.6 g of adipic acid dihydrazide was added as a crosslinker.

#### NAVP 9

25 A not autoxidisable crosslinkable urethane-acrylic hybrid bearing carbonyl functional groups in the acrylic polymer

##### Stage 1 - Synthesis of a polyurethane prepolymer

An isocyanate terminated urethane prepolymer was prepared as described below: 632 g IPDI (isophorone diisocyanate), 88 g dimethylol propanoic acid (DMPA), 880 g of a polypropylene glycol diol with an average molecular weight of 1600 g/mole (OH number = 77.4 mg KOH/g) and 4 drops of Dabco T-9 (a tin octoate catalyst from Air Products) were added to a 2 litre 3 necked round bottom flask. The mixture was stirred under nitrogen and heated to 95°C and kept at this temperature until the practical NCO content reached 7.7% (g NCO/ g prepolymer). The batch was cooled down to 70°C. 54.8 g TEA was added to the prepolymer and the mixture was kept at this temperature for 1 hour.

##### Stage 2 - Preparation of a polyurethane dispersion from the polyurethane prepolymer prepared in Stage 1

650 g of prepolymer prepared in Stage 1 was added to a water phase consisting of 977.9 g demineralised water and 16.25 g of the surfactant Synperonic NP9 (available from ICI Surfactants). The total addition time was 70 minutes. When the addition was complete, a mixture of 27.0 g of hydrazine hydrate (63.5%) and 27 g water was added to the dispersion. The temperature of the dispersion increased to 41°C and was kept at this temperature for 30 minutes. The batch was cooled down to room temperature and sieved through a 200 mesh sieving cloth. The dispersion had a solids content of 39%, pH was 7.4 and viscosity at 25°C was 160 mPa.s.

Stage 3 - Synthesis of a non-oxidisably crosslinkable urethane-acrylic hybrid bearing carbonyl functional groups from the polyurethane dispersion of stage 2

55.6 g butyl methacrylate, 12.2 g methyl methacrylate, 10.1 g diacetone acrylamide and 130 g demineralised water were added to 400 g of polyurethane dispersion from stage 2 and slowly heated to 36°C. Then 0.35 g of a Fe(II)EDTA solution (1% w/w in water), 5.1 g of a tertiary butyl hydroperoxide solution (4.7% w/w in water) and 22.6 g iso-ascorbic acid solution (1% w/w in water) was added. The temperature was increased to 48°C and was kept on this temperature for 18 minutes and cooled to room temperature. 55.6 g butyl methacrylate, 12.2 g methyl methacrylate, 10.1 g diacetone acrylamide and 130 g demineralised water were subsequently added and slowly heated to 31°C and 22.6g iso-ascorbic acid solution (1% w/w in water) was added. The temperature was increased to 35°C and 11.3g iso-ascorbic acid solution (1% w/w in water) was added and kept on this temperature for 13 minutes and again 11.3g iso-ascorbic acid solution (1% w/w in water) was added and temperature was kept on 32 °C for 1 hour. The resulting urethane-acrylic hybrid dispersion was cooled down to room temperature and sieved through a 200 mesh sieving cloth. The dispersion had a solids content of 35.2%, pH was 7.8 and viscosity at 25°C was 40 mPa.s. To 400 g of this dispersion 3.3g of adipic acid dihydrazide was added as a crosslinker.

Comparative vinyl-polymers

Preparation of not oxidisably crosslinkable vinyl polymers not bearing carbonyl functional groups = CVP's 10 to 12

CVP 10

A not autoxidisably crosslinkable acrylic dispersion not bearing carbonyl functional groups

911.86 g of demineralised water and 1.82 g of sodium lauryl sulphate (30% w/w) were added to a three necked round bottom flask, equipped with a thermometer, stirrer, cooler and a N<sub>2</sub> inlet. The mixture was heated to 70°C under N<sub>2</sub> atmosphere.

5 % of a monomer mixture consisting of 188.2 g of demineralised water, 5.45 g of sodium lauryl sulphate 30% (w/w), 6.98 g of lauryl mercaptane, 3.49 g of mercapto

propionic acid, 43.61 g of methacrylic acid and 392.44 g of methylmethacrylate was added to the reactor. The reactor content was then heated to 80°C.

30% of a solution of ammonium persulphate (87.2g, 1.5% w/w) in water was added to the reactor. The reactor content was heated to 85°C and mixed at this temperature for 5 minutes.

The remainder of the monomer mixture was added to the reactor after 60 minutes, the remainder of the ammonium persulphate solution was added to the reactor after 70 minutes. The reactor content was kept at 85°C during this period. After the addition was complete, the reactor content was held at 85°C for another 30 minutes.

A solution of 34.50 g of ammonia (25% w/w) and 124.4 g of demineralised water was added to the reactor. The reactor content was held at 85°C for another 20 minutes. After this the batch was cooled down to room temperature and sieved through a 200 mesh sieve cloth. The pH was 8.3 and the viscosity was 12 mPas at 25°C.

1055.60g of this polymer dispersion was mixed with 184.15 g of demineralised water for 15 minutes at room temperature in the same type of three necked round bottom flask. 0.22 g of dimethylethanol amine and 170.34 g of butylmethacrylate and 46.55 g of butylacrylate were added to the reactor, followed by 1.88 g of tert.-butylhydroperoxide (30% solution in water w/w) and 2.86 g of a 1% (w/w) solution of iron ethylenediaminetetraacetate (FeEDTA) in water. The reactor content was heated to 35°C and mixed for 60 minutes.

2.75g of a 3.96% (w/w) solution of l-ascorbic acid in water was added and the reactor content was heated to 57°C in 25 minutes. The batch was held at this temperature for 15 minutes.

Then 37.75 g of a 1.21% (w/w) solution of l-ascorbic acid in water was added to the reactor in 30 minutes at 57°C. The batch was mixed for 15 minutes and cooled down to 35°C.

A mixture of dimethyl ethanol amine (0.22 g), butyl methacrylate (170.4 g) and butyl acrylate (46.5 g) was added to the reactor. 1.88 g of tert.-butylhydroperoxide (30% solution in water w/w), 2.86 g of a 1% (w/w) solution of FeEDTA in water and 235.77 g of demineralised water were added to the reactor and mixed for 60 minutes at 35°C. 2.75 g of a 3.96% (w/w) solution of l-ascorbic acid in water was added to the reactor content. The batch was heated to 42°C and mixed at this temperature for 15 minutes.

Then 37.75 g of a 1.21% (w/w) solution of l-ascorbic acid in water was added to the reactor in 30 minutes at 42°C.

The batch was mixed for 30 minutes at 40°C, cooled down to room temperature and sieved through a 200 mesh sieve cloth.

The specifications of this batch were as follows: solids content 35%, pH of 7.9 and viscosity of 24 mPas at 25°C.

**CVP 11**A not autoxidisable crosslinkable acrylic dispersion not bearing carbonyl functional groups

859.8g of demineralised water, 14.0g of sodium lauryl sulphate (30w/w), 1.69g ammonium persulphate and 1.03g of ammonium bicarbonate (25% w/w) were added to a three necked round bottom flask, equipped with a thermometer, stirrer, cooler and a N<sub>2</sub> inlet. 10% of a monomer mixture consisting of 242.0g of demineralised water, 28.1 g of sodium lauryl sulphate (30% w/w), 2.53g ammonium persulphate, 2 g ammonium bicarbonate (25% w/w), 33.7g methacrylic acid, 421.2g butylmethacrylate, 166.1g butylacrylate and 221.4g styrene was added to the reactor. This mixture was heated to 80°C under N<sub>2</sub> atmosphere and mixed at this temperature for 20 minutes. The remainder of the monomer mixture was added to the reactor during 90 minutes. The reactor content was kept at 80°C during this period. After the addition was complete, the reactor content was held at 80°C for another 30 minutes. A solution of 3.08 g of ammonia (25% w/w) and 3.34g of demineralised water was added to the reactor. The reactor content was held at 80°C for another 30 minutes. After this the batch was cooled down to room temperature and sieved through a 200 mesh sieve cloth. The pH was 6.5 and the viscosity was 36 mPa.s

**CVP 12**A not autoxidisable crosslinkable urethane-acrylic hybrid dispersion not bearing carbonyl functional groups

66.6g butyl methacrylate, 11.5g methyl methacrylate and 130g demineralised water were added to 400g of polyurethane dispersion from stage 2 from Example 10 and slowly heated to 35°C. Then 0.35g of a Fe(II)EDTA solution (1% w/w in water), 5.1g of a tertiary butyl hydroperoxide solution (4.7% w/w in water) and 22.6g iso-ascorbic acid solution (1% w/w in water) was added. The temperature was increased to 44°C and was kept on this temperature for 18 minutes and cooled to room temperature. 66.6g butyl methacrylate, 11.5g methyl methacrylate and 130g demineralised water were subsequently added and slowly heated to 34 °C and 22.6 g iso-ascorbic acid solution (1% w/w in water) was added. 11.3g iso-ascorbic acid solution (1% w/w in water) was added and kept on this temperature for 13 minutes and again 11.3g iso-ascorbic acid solution (1% w/w in water) was added and temperature was kept on 33 °C for 1 hour. The resulting urethane-acrylic hybrid dispersion was cooled down to room temperature and sieved through a 200 mesh sieving cloth. The dispersion had a solids content of 35%, pH was 8.2 and the viscosity at 25°C was 50 mPa.s at 25°C.

Examples 1 to 30 and Comparative Examples C1 to C25

Preparation of compositions according to the invention were carried out by blending at room temperature the autoxidisable crosslinkable organic polymers containing

unsaturated fatty acid residues [FATTY ACIDS 1 to 5] with the not autoxidisably crosslinkable vinyl polymers bearing carbonyl functional groups [NAVp's 6 to 9].

Comparative examples C1 to C25 were prepared by blending the autoxidisably crosslinkable organic polymers containing unsaturated fatty acid residues [FATTY ACIDS 1 to 5] with not autoxidisably crosslinkable vinyl polymers not bearing carbonyl functional groups [CVP's 10 to 12], at room temperature.

The composition and the properties of resultant coatings are shown below in Tables 1 to 6.

i) Resistance. The blends were cast down on Leneta test charts Form 2C with a film thickness of 100 µm. The films were dried at room temperature for 20 minutes and at 50°C for 64 hours. After they were cooled down to room temperature the films were tested for "household chemical resistance".

a) Drops of the various testing liquids (water, ethanol, coffee and "Andy" a commonly used Dutch detergent) were placed on the films and covered with a watch glass. The liquids were removed after 16 hours at room temperature and the damage to the coating was assessed immediately and after four hours recovery. 0 means that the coating was dissolved, 5 means that the coating was not affected at all.

For the hot pan test a drop of boiling water was placed on the film and covered with a beaker containing boiling water for one hour before removing the beaker and immediately assessing the film. 0 means that the coating was dissolved, 5 means that the coating was not affected at all.

ii) BHMR = Black heel mark resistance: the coating is hit with a black heel and the degree of damage to the coating was assessed the same as the household chemical resistance is assessed.

iii) Yellowness (Yellow): 0 is very yellow, 5 is a colourless coating.

iv) Pencil hardness: was assessed according to ASTM D3363-92a. H numbers are hard and B numbers are soft on a scale from 8H to HB to B or less.

v) Hardness development: The Hardness was measured using a Pendulum Hardness Tester according to the König test method. The larger the value the harder the film was.

Examples 1 to 2: Blends of FATTY ACID 1 with NAVP 6 and  
Comparative Examples C1 to C5: FATTY ACID 1, NAVP 6, CVP10 and  
Blends of FATTY ACID 1 and CVP 10

5

Table 1

Example	vinyl polymer	FATTY ACID 1/ vinyl polymer ratio (w/w)	BHMR	Yellow	Stain resistances (assessment after 4 hrs)							
					Water		EtOH		Coffee		Andy	
C1	none	100/0	5	0	5*	5	5*	5	4	4	5*	5
C2	CVP 10	0/100	1	5	5*	5	2*	2	4	4	1*	1
C3	NAVP 6	0/100	1	5	5*	5	4*	4	5	5	4*	4
C4	CVP 10	50/50	2	2	5*	5	1*	1	4	4	2*	3
1	NAVP 6	50/50	3	2	5*	5	2*	2	5	5	4*	5
C5	CVP 10	25/75	1	0	5*	5	1*	1	5	5	0*	0
2	NAVP 6	25/75	4	0	5*	5	2*	2	5	5	5	5

\* = immediate assessment



Examples 3 to 8: Blends of FATTY ACID 1 with NAVP 7, 8 and 9  
Comparative Examples C8 to C9: Blends of FATTY ACID 1 with CVP 11 and 12

Table 2

Example	vinyl polymer	FATTY ACID 1/ vinyl polymer ratio (wt/wt)	BHR	Yellow	Pencil Hardness	Stain resistances (assessment after 4 hrs)				Hardness Development (room temperature)			
						Water	EtOH	Coffee	Andy	3 hrs	6 hrs	1 day	4 days
3	NAVP 7	25/75	3-4	-	5H	5	5	4	4-5	27	45	87	99
4	NAVP 7	50/50	1	-	7H	5	4	3-4	5	30	41	67	86
5	NAVP 8	25/75	2	-	5H	5	4-5	4	5	30	30	83	89
6	NAVP 8	50/50	0	-	6H	5	4	3	5	27	39	71	73
C8	CVP 11	25/75	0	-	5H	4-5	0	4	3-4	24	37	64	74
C7	CVP 11	50/50	0	-	5H	5	2	3-4	2-3	28	39	67	79
7	NAVP 9	50/50	3-4	4-5	7H	5	4	1	5	50	61	80	81
8	NAVP 9	25/75	2	5	6H	5	3	1	5	76	84	97	91
C8	CVP 12	50/50	1	2-3	5H	5	4	1	1	53	70	80	84
C9	CVP 12	25/75	0	2-3	7H	4-5	4	2	1	61	73	77	80

- = not measured

Example 9 to 14: Blends of FATTY ACID 2 with NAVP 7, 8 and 9  
and Comparative Examples C10 to C13: Blends of FATTY ACID 2 with CVP 11 and 12

Table 3

Example	vinyl polymer	FATTY ACID 2/ vinyl polymer ratio (w/w)	BHMR	Pencil Hardness	Stain resistances (assessment after 4 hrs )				Hardness Development (room temperature)				
					Water	EtOH	Coffee	Andy	3 hrs	6 hrs	1 day	4 days	
9	NAVP 7	50/50	3	6H	4	3-4	2-3	4	11	11	37	-	
10	NAVP 7	25/75	0-1	5H	3-4	3	3	4-5	14	14	61	-	
11	NAVP 8	50/50	2	7H	3-4	1	2-3	4	11	14	44	-	
12	NAVP 8	25/75	0	7H	3-4	3	3	4-5	10	17	54	-	
C10	CVP 11	50/50	0	5H	1-2	1	2-3	3	lacky	13	37	-	
C11	CVP 11	25/75	0	5H	1-2	1	3-4	3-4	lacky	17	57	-	
13	NAVP 9	50/50	0	5H	4	4	1	4-5	27	31	48	53	
14	NAVP 8	25/75	1	8H	5	4	1	5	43	50	59	64	
C12	CVP 12	50/50	0	B	2	5	1	4	11	33	40	40	
C13	CVP 12	25/75	0	6H	5	3	1	3	31	43	50	51	

- = not measured

Examples 15 to 18 - Blends of FATTY ACID 3 with NAVP 7 and 9 and  
Comparative Examples C14 to C17: Blends of FATTY ACID 3 with CVP 11 and 12

Table 4

Example	vinyl polymer	FATTY ACID 3/ vinyl polymer ratio (w/w)	BHMR	Yellow	Pencil Hardness	Stain resistances (assessment after 4 hrs)				Hardness Development (room temperature)			
						Water	EtOH	Coffee	Andy	3 hrs	6 hrs	1 day	4 days
15	NAVP 7	25/75	2	-	<B	5	1	3	3	7	8	17	17
16	NAVP 7	50/50	0	-	<B	2	1	1	1	lacky	lacky	lacky	9
C14	CVP 11	25/75	0	-	<B	3	1	3	3	6	6	6	13
C15	CVP 11	50/50	0	-	<B	1	1	1	1	lacky	lacky	lacky	6
17	NAVP 9	50/50	0	1-2	<B	3	3	0-1	0	11	11	11	11
18	NAVP 9	25/75	1-2	2-3	<B	4-5	4	1	4	16	17	17	17
C16	CVP 12	50/50	0	1-2	<B	2	3	1	0	11	11	11	11
C17	CVP 12	25/75	0	1-2	<B	2	4	1	2	21	20	20	20

- = not measured

Examples 19-24: Blends of FATTY ACID 4 with NAVP 7, 8 and 9 and Comparative Examples C18 to C21 Blends of FATTY ACID 4 with CVP 11 and 12

Table 5

Example	vinyl polymer	FATTY ACID 4/ vinyl polymer ratio (w/w)	BHMR	Yellow	Pencil Hardness	Slain resistances (assessment after 4 hrs)				Hardness Development (room temperature)			
						hot pan	EIOH	Coffee	Andy	4 hrs	1 day	2 days	
19	NAVP 7	50/50	4	-	-	4	4-5	4-5	4-5	19	54	63	-
20	NAVP 7	25/75	3	-	-	2	4-5	4	4-5	17	50	63	-
21	NAVP 8	50/50	3	-	-	2	4-5	4-5	4-5	20	47	59	-
22	NAVP 8	25/75	0	-	-	4	4	4	4-5	11	24	39	-
C18	CVP 11	50/50	0	-	-	4	2	2	2	13	43	47	-
C19	CVP 11	25/75	0	-	-	3-4	3	3	2	13	39	41	-
						Water	EIOH	Coffee	Andy	3 hrs	6 hrs	1 day	4 days
23	NAVP 9	50/50	3-4	2-3	7H	5	4	1	4-5	49	59	81	90
24	NAVP 9	25/75	3-4	4-5	7H	5	4	1	5	87	94	104	110
C20	CVP 12	50/50	0	1-2	7H	5	4-5	2	3-4	41	63	70	79
C21	CVP 12	25/75	0	2-3	6H	5	4-5	1	3	69	73	74	80

- = not measured

Examples 25 to 30: Blends of FATTY ACID 5 with NAVP 7, 8 and 9 and  
Comparative Examples C22 to C25: Blends of FATTY ACID 5 with CVP 11 and 12

Table 6

Example	vinyl polymer	FATTY ACID 5/ vinyl polymer ratio (wt/wt)	BHMR	Pencil Hardness	Stain resistances (assessment after 4 hrs )			Hardness Development (room temperature)				
					Water	EtOH	Coffee	Andy	3 hrs	6 hrs	1 day	4 days
25	NAVP 7	50/50	0	6H	-	4	3-4	2	23	-	80	91
26	NAVP 7	25/75	0	6H	-	3	4-5	3	19	-	63	86
27	NAVP 8	50/50	0	6H	-	3	3-4	3	17	-	64	80
28	NAVP 8	25/75	0	5H	-	3-4	4-5	3	19	-	68	91
C22	CVP 11	50/50	0	4H	-	2	4	2-3	19	-	64	73
C23	CVP 11	25/75	0	5H	-	1	4	2	14	-	67	89
29	NAVP 8	50/50	2	8H	5	4	1	4	64	77	91	90
31	NAVP 9	25/75	2	8H	5	4-5	1	4-5	83	91	103	103
C24	CVP 12	50/50	0	7H	5	4	1	2	40	56	68	67
C25	CVP 12	25/74	0	7H	5	4	1	1	64	74	79	77

- = not measured

CLAIMS

1. An aqueous crosslinkable coating composition comprising as aqueous dispersed components:
  - 5 (i) at least one autoxidisable crosslinkable organic polymer containing unsaturated fatty acid residues, and
  - (ii) at least one vinyl polymer which is not autoxidisable crosslinkable and bears carbonyl functional groups, andwherein said composition has present therein carbonyl reactive amine and/or hydrazine  
10 functional groups which impart crosslinkability to component (ii).
2. A composition according to claim 1 wherein component (i) is a polyurethane polymer.
- 15 3. A composition according to claim 2 wherein the polyurethane polymer has been formed by the chain extension of an isocyanate-terminated polyurethane prepolymer with an active hydrogen chain extending compound.
- 20 4. A composition according to any one of the preceding claims wherein the functional amine or hydrazine groups are provided by polyamines or polyhydrazines which are not part of the autoxidisable crosslinkable organic polymer or of the vinyl polymer.
- 25 5. A composition according to any one of the preceding claims wherein the hydrazine functional groups are provided by compounds selected from dicarboxylic acid bis-hydrazides and hydrazones derived therefrom, carbonic acid hydrazides and bis-semicarbazides.
- 30 6. A composition according to any one of the preceding claims wherein the autoxidisable crosslinkable organic polymer containing unsaturated fatty acid residues also bears chain-pendant carbonyl functional groups.
7. A composition according to any one of the preceding claims wherein the autoxidisable crosslinkable polymer containing unsaturated fatty acid residues also bears  
35 chain-pendant amine or hydrazine derivative functional groups.
8. A composition according to any one of the preceding claims wherein the vinyl polymer also bears chain pendant amine or hydrazine derivative functional groups.

9. A composition according to any one of the preceding claims wherein the ratio of amine or hydrazine derivative groups to carbonyl functional groups is 0.02 to 1.6 moles of amine or hydrazine derivative groups per mole of carbonyl functional groups.

5 10. A composition according to any one of the preceding claims wherein the ratio of component i) to component ii) is from 90:10 to 10:90.

11. A coated substrate having a coating obtainable from an aqueous crosslinkable coating composition as claimed in any one of claims 1 to 10.

10

12. Use of an aqueous crosslinkable coating composition as claimed in any one of claims 1 to 10 for coating a substrate.

# INTERNATIONAL SEARCH REPORT

Int. Application No PCT/GB 00/01011	
<b>A. CLASSIFICATION OF SUBJECT MATTER</b> IPC 7 C09D175/06 C09D133/24 C08G18/08 C08G18/42	
According to International Patent Classification (IPC) or to both national classification and IPC	
<b>B. FIELDS SEARCHED</b> Minimum documentation searched (classification system followed by classification symbols) IPC 7 C09D C08G	
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched	
Electronic data base consulted during the international search (name of data base and, where practical, search terms used)	
<b>C. DOCUMENTS CONSIDERED TO BE RELEVANT</b>	
Category *	Citation of document, with indication, where appropriate, of the relevant passages Relevant to claim No.
A	DATABASE WPI Section Ch. Week 199805 Derwent Publications Ltd., London, GB; Class A21, AN 1998-046982 XP002140913 & JP 09 296023 A (NIPPON SHOKUBAI CO LTD), 18 November 1997 (1997-11-18) abstract
A	PATENT ABSTRACTS OF JAPAN vol. 018, no. 182 (C-1184), 29 March 1994 (1994-03-29) & JP 05 339542 A (KANSAI PAINT CO LTD), 21 December 1993 (1993-12-21) abstract
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<input checked="" type="checkbox"/> Further documents are listed in the continuation of box C.	
<input checked="" type="checkbox"/> Patent family members are listed in annex.	
* Special categories of cited documents : "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "Z" document member of the same patent family	
Date of the actual completion of the international search  23 June 2000	Date of mailing of the international search report  06/07/2000
Name and mailing address of the ISA European Patent Office, P.B. 5618 Patentaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-3040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016	Authorized officer  Neugebauer, U



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Int'l. Patent Application No.  
PCT/GB 00/01011

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A	DE 43 28 092 A (HERBERTS GMBH) 23 February 1995 (1995-02-23) page 2, line 16 -page 4, line 19 example 10 -----	1, 3, 11, 12

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